APPENDIX A

CASE STUDY OF THE NEW YORK METROPOLITAN REGION

In order to give concreteness to the investigation, the oxidant problem of the New York metropolitan region has been examined in further detail. 20 The entire Northeast has a severe problem with photochemical smog. The national ambient air quality standard for ozone is violated for virtually the entire Northeast region. Summer peak oxidant levels downwind of New York attain levels four to six times the primary standard of.12 In part, this is the result of the high concentration of mqq. population and industry in the area. It has been estimated that over fifty percent of all man-made HC and $\mathtt{NO}_{\mathbf{x}}$ emissions are generated in the Northeast region. Transport is also critically important. Prevailing summer winds from the southwest are heavily laden with oxidants and precursors from the mid-Atlantic industrial belt, and carry pollutants from the New York metropolitan region across western Connecticut and Massachusetts, sometimes as far north as southern New Hampshire and Maine. Violations of ambient air quality standards in rural New England area are often entirely attributable to transported materials. 21 $\mathtt{NO}_{\mathbf{x}}$ emissions from the New York area are also implicated in the formation of nitric acid and the acidification of precipitation in New England and Canada.

Oxidant control strategy in the New York metropolitan region has concentrated almost entirely on the abatement of hydrocarbon emission. For example, the 1979 revisions to State

Implementation Plans, required as policy changes to bring about attainment of national standards, imposed new controls only on hydrocarbon sources. Reasonably Available Control Technologies

(RACT) were imposed on fifteen major stationary sources of HC emissions. Vehicle Inspection and Maintenance Programs were either installed or strengthened throughout the region as additional controls on mobile HC emission sources. This policy has been sustained, despite scientific warnings that due to the superabundance of hydrocarbons in the atmosphere, ozone levels seem insensitive to such abatement efforts, 23

Despite substantial control efforts, there has been little long-term improvement in oxidant measurements in the metropolitan region. Moreover, simulations attempting to estimate the consequences of recent additional abatement measures adopted in 1979, as well as further possible steps aimed mainly at hydrocarbon emissions, indicate that there is little prospect that the region will come into compliance with national standards in the forseeable future.²⁴

Isocost-Abatement-Possibilities

In New York, larger reductions in current emissions levels would be required to approach national air quality standards. These cuts would go far beyond those available at small cost into the realm of abatement measures with high and rapidly increasing incremental costs. Consideration of the behavior of these costs is crucial in policy analyses.

Two kinds of information contribute to the derivation of isocost abatement possibilities. The first concerns available control measures relevant to specific kinds of emission sources of hydrocarbons and NO_{χ} , and their cost-effectiveness, in terms of dollars per ton of material removed. Typically, there will be several control options for each emissions source. These will differ both in control efficiency (percentage of uncontrolled emissions removed) and cost. These control options for all sources of a pollutant can be arrayed from cheapest (in cost per ton abated) to most expensive.

In order to derive incremental and total abatement cost schedules from these data, the distribution, or inventory, of emissions by source category must also be known. Such an inventory consists of estimates, in tons per year, of emissions from a disaggregated list of source types that correspond to the categories for which control costs are available. Such an inventory has been compiled for the New York Metropolitan region. There are considerable elements of projection and estimation in these inventories. Even large stationary sources have not been

accurately enumerated; smaller ("area") sources are estimated indirectly, through data on population, employment, fuel consumption and other means. Rarely are emissions actually Typically, they are estimated through the use of measured. "emissions factors", which represent average or otherwise representative emissions rates for specific types and sizes of sources. Further, baseline inventories incorporate the projected future effects of past emission control regulations that have not yet fully been reflected in actual emissions levels. An important example is the baseline projection of vehicle emissions, which depends on the scrapping rate of older cars not subject to the federal motor vehicle emissions standards, the growth of new car registrations, the rate of deterioration of emissions control efficiencies as cars age, and changes in traffic volumes and patterns.

The baseline inventories for HC and NO_X are presented in Table A-1 and A-2 respectively. 25 These are intended to reflect changes in emissions resulting from regulations promulgated prior to 1979, but not from the changes made in the 1979 revisions to State Implementation Plans, because those changes are included among the abatement possibilities for which cost-effectiveness calculations have been made. The exception to this distinction in the final reduction in NO_X vehicular emissions standards to 1.0 grams per mile. Although this reduction was mandated in the 1977 Clean Air Act Amendments, its implementation has been .PA

Table A-1. New York Metropolitan Region: Baseline Hydrocarbons Emissions Inventory

Source Category	Tons/Year
Vehicle Surface Coating Metal Can Coating Coil Coating Wire Coating Fabric Coating Paper Coating Metal Furniture Coating Miscellaneous Metal Parts Coating Flat Wood Coating Manufacture of Synthetic Pharmaceuticals Manufacture of Rubber Tires Graphic Arts Processes Dry Cleaning Degreasing Operations Miscellaneous Refinery Losses Fixed Roof HC Storage Tanks Floating Roof HC Storage Tanks Petroleum Bulk Plants Gas Terminals Gas Stations Fuel Combustion Other Stationary Sources	13,831 8,705 153 179 14,900 6,110 90 26,000 94 13,700 1,570 3,800 4,200 9,883 4,540 8,270 1,800 68,495 124,537 47,464 25,031 212,117
SUBTOTAL	610,458
Vehicle Emissions	164,922
TOTAL	730,391

Table A-2. New York Metropolitan Region: Baseline $\mathrm{NO}_{\scriptscriptstyle x}$ Emissions Inventory

Area & Mobile Sources t/y	Tons/Year
Residential Coal Residential Oil Distillate Oil Natural Gas L.P.G. Wood/Other	190 1,234 17,459 7,459 122
Commercial Coal Residential Oil Distillate Oil Natural Gas L.P.G.	22,071 4,793 3,825
Wood/Other Industrial Coal Residential Oil Distillate Oil Natural Gas L.P.G.	 7,719 13,076 4,155
Wood/Other Aircraft Railroad Vessels Off-Highway Vehicles (gas, diesel) Burning	10,834 7,330 15,920 45,635 3,505
TOTAL AREA	165,100
Light-duty Autos Trucks Heavy-duty Gas Diesel Motorcycles	285,180 39,321 23,418 29,944 2,351
TOTAL MOBILE	380,214

Table A-2. (Continued)

Boilers, electrical generating Boilers, industrial Boilers, comm./inst. Space Heaters, industrial Space Heaters, comm./inst. Other fuel combustion Solid waste/govt. Solid comm./inst. Solid End. Waste Solvent Recovery Stationary, i.e., engines N.C.R.	314,605 41,387 21,509 7 3 455 812 77 154 2,335 306
TOTAL POINT	381,650

postponed and is still not completely certain. The costs of successive reductions from 3.0 to 2.0 and from 2.0 to 1.0 grams per mile have been included in the ${
m NO}_{
m X}$ incremental abatement cost calculations.

It is clear that the sources of $\mathbf{NO}_{\mathbf{X}}$ emissions are more concentrated than those of hydrocarbons. Fuel combustion, mainly in large industrial and electrical utility boilers, accounts for by far the major part of stationary source emissions. attendant on industrial processes are relatively unimportant. Most of the remaining $\mathtt{NO}_{\mathbf{X}}$ emissions stem from internal combustion gasoline or diesel engines, predominantly vehicular. The industrial sources of hydrocarbon emissions are more varied. Substantial evaporative losses to the atmosphere occur during coating operations using petroleum-based solvents, and in degreasing operations using solvents. The bulk of HC losses are transportation-related however, either through direct vehicle emissions or through losses that occur while sorting, transporting, or transferring petroleum products. For both HC and $\mathbf{NO}_{\mathbf{x}}$, substantial aggregate emissions are attributable to a variety of relatively small, "area sources" and estimated with considerable imprecision.

Complementary to these data is information on the possibilities for abatement of emissions, category by category, and the associated resources costs. Estimates of control efficiencies and costs are very approximate, and their inaccuracy reflects the inherent limitations of the regulatory processes. This information is rarely tendered by regulated sources, for whom such data are internal, if not confidential. Estimates are

made by the regulatory agencies or their consultants, and invariably represent conditions facing average, representative or typical sources within a category, since efforts to estimate cost-effectiveness for the "typical" example may be quite inaccurate even when categories are narrowly defined. The actual data available refer to rather broad categories as Table A-1 and A-2 demonstrate. The estimates, incremental abatement cost curves and isocost curves derived from them, should thus be regarded as approximate figures indicating general tendencies.

Incremental costs schedules for hydrocarbon and NO_X abatement are presented in Tables A-3 and A-4, respectively.26 Each control measure has two dimensions: an estimated cost per ton abated, and an estimate of the tonnages that can be controlled by the measure. The costs include annualized capital costs and annual operating, maintenance and other costs, including credits for materials recovery and penalties for energy losses.

Both incremental cost schedules share two distinctive features. The first is that some emissions can be abated at negative cost. Substantial tonnages of hydrocarbons emissions can be prevented at actual cost savings, because for evaporative losses from bulk storage facilities and other high-volume operations, the value of the hydrocarbon material recovered would exceed the cost of the equipment needed for recovery. Savings from abatement of ${\rm NO}_{\rm X}$ emissions from fuel combustion arise from improvements in burner efficiency. The second distinctive

Table A-3. Incremental Cost Schedule for Hydrocarbon Abatement in the NY Metropolitan Region

Source Category	Cost (\$/ton)	Tons/Year
petroleum tank truck terminals cutback asphalt dry cleaning refinery leakages misc. refinery sources degreasing operations gasoline bulk storage rotogravure & printing upgraded vehicle I & M (NJ) automobile coating paper coating fabric coating coil coating can coating wire coating flat wood coating gas stations misc. metal parts coating metal furniture coating can coating coil coating can coating metal furniture (CT) auto coating new I & M program (NY) intensified I & M, regional auto coating bulk storage advanced I & M, regional	-273 -128 -111 -107 - 88 - 80 14 18 30 38 53 59 91 108 137 331 359 563 754 1016 1100 1139 1428 1720 2653 4150 5800 5982	123,292 8,214 2,110 1,777 2,033 2,965 55,481 2,470 15,384 11,203 5,988 14,155 138 7,834 170 82 23,732 20,800 81 783 7 17,562 8,061 1,245 4,728 3,180 691 12,329 3,817

Table A-4. Incremental Cost Schedule for $\mathrm{NO}_{_{\mbox{\scriptsize X}}}$ Abatement in the NY Region

Source Category	Abatement Level or Method	Cost \$/t	Tons/ Year
industrial boilers,			
residual oil industrial boilers,	LEA	-2306	3,559
distillate oil	LEA	-1554	83
industrial boilers, natural gas	LEA	-1225	497
commercial boilers, natural gas	LEA	-1225	108
commercial boilers,			
distillate oil utility boilers,	LEA	-1000	387
oil & gas industrial process	LEA	- 487	45,303
furnaces utility boilers, coa		- 397 - 173	41 1,258
non-utility boilers, oil & gas utility boilers, coa	SCA	0 72	13,507 3,775
industrial boilers, natural gas	RAP	87	248
industrial process furnaces	ABD	101	41
heavy-duty diesel vehicles		152	22,458
stationary i.c. engines	SCR (dry)	245	2,101
heavy-duty gas vehicles	, a ₁ ,	343	17,563
utility boilers,	CCA	441	75,505
oil & gas non-utility boilers,			
natural gas utility boilers,	RAP & SCA	490	1,263
oil & gas industrial boilers,	SCR (PF)	508	60,404
residual oil light-duty vehicles	LNB EGR	665 785	7,119 94,109
industrial boilers, residual oil	NH ₃ Inj.	890	1,780
industrial boilers, residual oil (area)	NH ₃ Inj.	980	3,859

Table A-4. (continued)

Source Category I	Abatement Level or Method	Cost \$/t	Tons/ Year
industrial process			
furnaces industrial process	SCR (dry)	1037	66
furnaces utility boilers,	RAP & FGR	1079	745
oil & gas utility boilers, coal	SCR (MB) SCR (dry)	1133 1230	30,202 6,292
industrial boilers, distillate oil	LNB	1398	166
light-duty vehicles, advanced I&M programs industrial boilers,	3	1400	15,280
residual oil industrial boilers,	SCR (MB)	1816	14,237
distillate oil commercial boilers,	FGR	1833	41
residual oil (area) utility boilers,	NH ₃ Inj.	1880	11,036
oil and gas light-duty vehicles,	NH ₃ Inj.	2250	60,404
advanced emissions controls		2276	87,860
non-utility boilers, natural gas	SCR (FPB)	2812	1,067
industrial boilers, distillate oil	RAP & LNB	2958	124
<pre>industrial area souro distillate oil commercial boilers,</pre>	LNB	3206	6,538
residual oil industrial boilers,	NH ₃ Inj.	3250	3,925
distillate oil light-duty trucks,	RAP & FGR	3484	41
advanced NO _x controls industrial area sourc	ees .	3520	9,830
distillate oil industrial area sourc	FGR	3564	654
natural gas industrial area sourc	RAP & LNB	3650	2,908
natural gas commercial area sourc	RAP & FGR	4243	208
natural gas commercial area sourc	RAP & LNB	4460	2,678
distillate oil	LNB	6194	2,396

Table A-4. (continued)

Source Category	Abatement Level or Method	Cost \$/t	Tons/ Year
commercial boilers, distillate oil industrial boilers, distillate oil transport controls commercial boilers, residual oil commercial boilers,	FGR SCR (FRB) SCR(PF)	7333 10117 17200 20434	1,742 124 1,670 6,281
distillate oil	SCR (PFB)	40857	1,355

Control Device Abbreviations and Codes

	Codes
LEA = Low Excess Air	01
SCA = Staged Combustion Air	02
LNB = Low NO_{x} Burners	03
NH, Inj. = Ammonia Injection	04
SCR (PF) = Selective Catalytic Reduction	
(Parallel Flow)	05
SCR (MB) = Selective Catalytic Reduction	
(Moving Bed)	06
FGR = Flue Gas Recirculation	07
RAP = Reduced Air Preheat	80
RAP + FGR	09
RAP + LNB	10
SCR (FPB) = Selective Catalytic Reduction	
(Fixed Pack Bed)	11
RAP + SCA	12
RAP + NH ₃ Inj.	13
Dry SCR	14
Water Injection	15
Fine Tuning and Changing the Air/Fuel Ratio	16
Advanced Design Burner	17
Chilled Absorption	18

feature of both incremental cost schedules is that the insert incremental costs mount very rapidly after "easy" abatement possibilities have been exhausted, and more technologically complex and capital-intensive abatement methods have to be employed.

These features give definite form to the isocost abatement possibilities functions. Examples of such functions for the New York metropolitan region are provided in Figure a-1. Isocost schedules bend outward from the axes, reflecting the cost savings that can be obtained from the initial increments of abatement of both pollutants. This makes it likely that it would be costeffective to control both precursors to some extent, but a mixed control strategy would still not necessarily be efficient, since reduced emissions, especially of NO_X, can sometimes increase ozone concentrations.

Isocost curves are also severely bowed out, at higher expenditure levels nearly squared off in shape, because marginal abatement costs rise so rapidly for both precursors. This feature also militates against a strategy of controlling only one precursor to reduce ozone concentrations. Finally, also because of rapidly increasing incremental costs, there are sharply diminishing returns to rising expenditures on precursor abatement. A substantial reduction in emissions can be obtained for zero net expenditures, because of the cost savings obtainable on initial abatement possibilities. However, expenditure of as much as \$100 million in annualized costs obtains relatively little additional abatement.

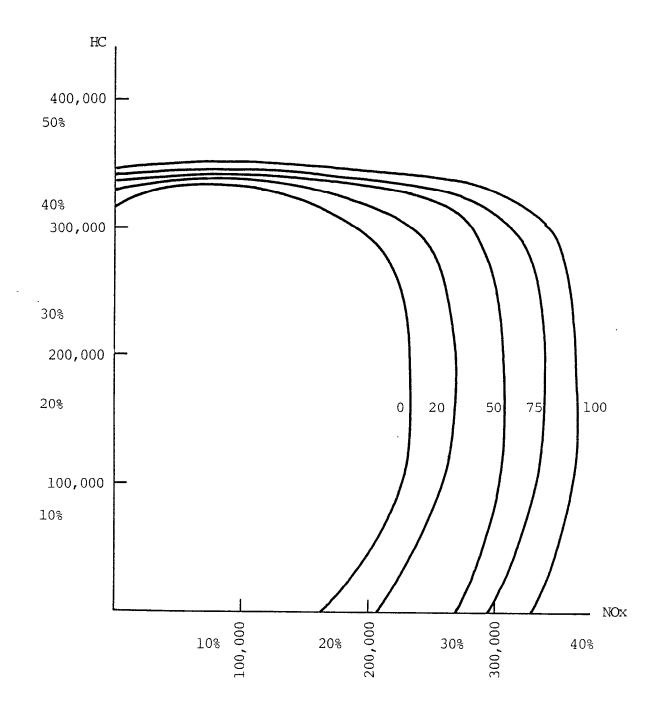


Figure A-1. Isocost Lines for Hydrocarbon and $\mathrm{NO}_{_{\mbox{\scriptsize X}}}$ Control

Isopleths of Precursor-Ozone Abatement Possibilities

The relative effects of NO_X and HC abatement on oxidant formation, as indicated by maximum ozone concentrations, were estimated through use of an EKMA (Empirirical Kinetic Modeling Approach) -- OZIPP (Ozone Isopleth Plotting Package) analysis for the entire region.²⁷ Because of the area-wide, relatively uniform photochemical oxidant problem throughout the region, ozone concentrations were predicted at only one location, represented by the monitory unit at Derby, Connecticut. OZIPP models a column of air containing initial concentrations of ozone and precursors moving along a fixed trajectory, picking up fresh emissions en route. Reactions occur in the column, depending on concentrations and solar energy, yielding predicted ozone levels at the design destination. EKMA then predicts the effects of specified percentage reductions in precursor concentrations.

Local conditions were used to construct isopleths specific to the New York region, as follows:

- 10 hour simulations, from 8 AM TO 6 PM;
- assumed reactivity of the nonmethan hydrocarbon mix equivalent to 25% propylene (ppmC), which is comparable to automobile exhaust;
- aldehydes at 5% of the initial NMHC mix;
- photolytic rate constants that vary in accordance with the diurnal variation in sunlight intensity on 7/21/77, the day the ozone design value was measured;
- an atmospheric dilution rate of 3%/hr. which lifted the mixing height from 500 m at 8 AM to 1300 m at 3 PM:
- daytime emission rates appropriate for a NY to Hartford trajectory;
- ozone aloft at 0.085 ppm, and negligible transported NO, and NMHC:

- normalized hourly variations in emission rates;
- exponential spatial decay of emission rates beyond the urban core.

Using the ozone isopleths thus derived, the EKMA procedure predicts future maximum hourly ozone concentrations after specified abatement of precursors. The procedure is calibrated using an ozone design value for the region of .323 ppm, the second highest concentration recorded at Derby, Connecticut and the ration of NMHC to NO_X emissions and 760,400 tons per year of hydrocarbon emissions, represents a 13% reduction of NO_X and a 25% reduction in the EKMA ozone design value. These reductions estimate the impacts of regulatory measures taken prior to the 1979 revisions of State Implementation Plans in the region, and for assumed growth rates and retirement rates in important sources of emissions.

The Nature of the Least-Cost Control Strategy

Figure a-2 superimposes the isopleths generated by the EKMA-OZIPP analysis for the New York metropolitan region, with emissions reductions expressed as percentage reductions n the baseline levels. Two isocost lines, representing total abatement expenditures of \$20 million and \$75 million annually are reproduced. The implications of nonconvexity in the damage function is immediately apparent in Figure a-2. Both the isocost schedules and the isopleths of ambient impact have the same general curvature. This means that the potential exists for "corner solutions," emphasizing only NO_X control or only HC control, to be cost effective. However, due to the strong

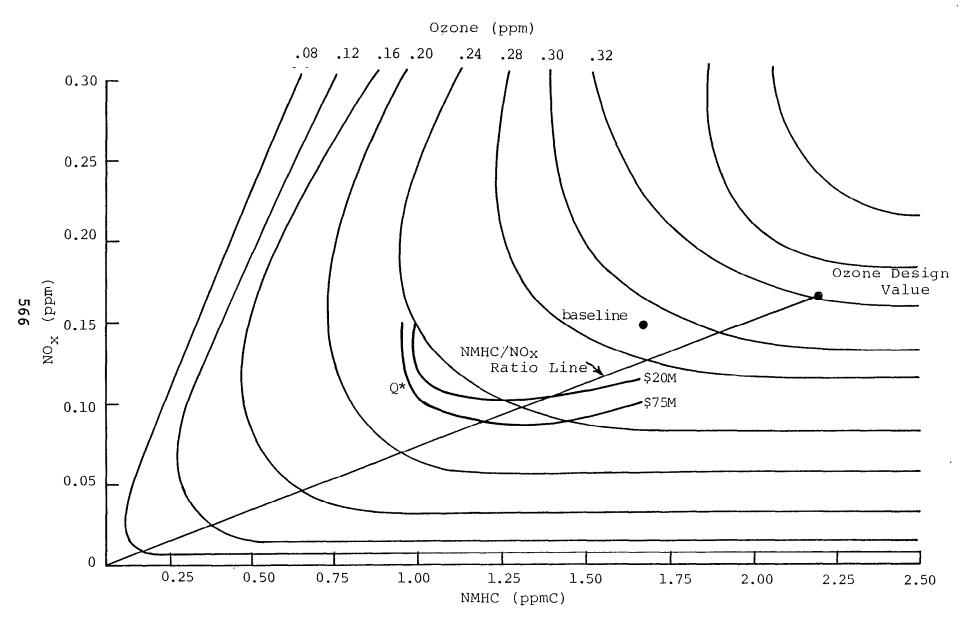


Figure A-2. Isopleth and Isocost Schedules: N.Y. Metropolitan Region

convexity in the isocost function, a reflection of sharply rising marginal control costs for both pollutants, it appears that a cost-effective control strategy in this case would still be to exercise some degree of control over both precursors. Point \mathbb{Q}^* , representing about a one-third reduction in baseline $\mathbb{NO}_{\mathbb{X}}$ emissions and a forty percent cut in hydrodynamic emissions, would be a cost-effective way of employing \$75 million for reducing peak ozone concentrations. It is to be noted, however, as other investigators have pointed out, that such an attainment and effort level would fall far short of that required to meet the national ambient air quality standard for ozone of .12 ppm.

This finding illustrates the need for attention both to effectiveness and cost parameters in exploring issues of costeffectiveness. Despite the strong nonconvexities, it appears that the inflexibilities of marginal costs still dictate, within the range of the New York data, at least, a mixed control strategy. However, Figure a-2 illustrates another implication of nonconvexity. The isocost line parallels the isopleth contours over a wide range of control combinations. This means that the marginal conditions for a least-cost strategy are nearly fulfilled by a wide range of abatement options. The least-cost strategy is thus not robust, in the sense that small changes in the cost or effectiveness parameters could imply large shifts in the efficient control strategy. Figure a-3 illustrates one such possibility. Since the net cost of hydrocarbon control in control options involving product recovered, a substantial increase in hydrocarbon prices would shift the isocost line

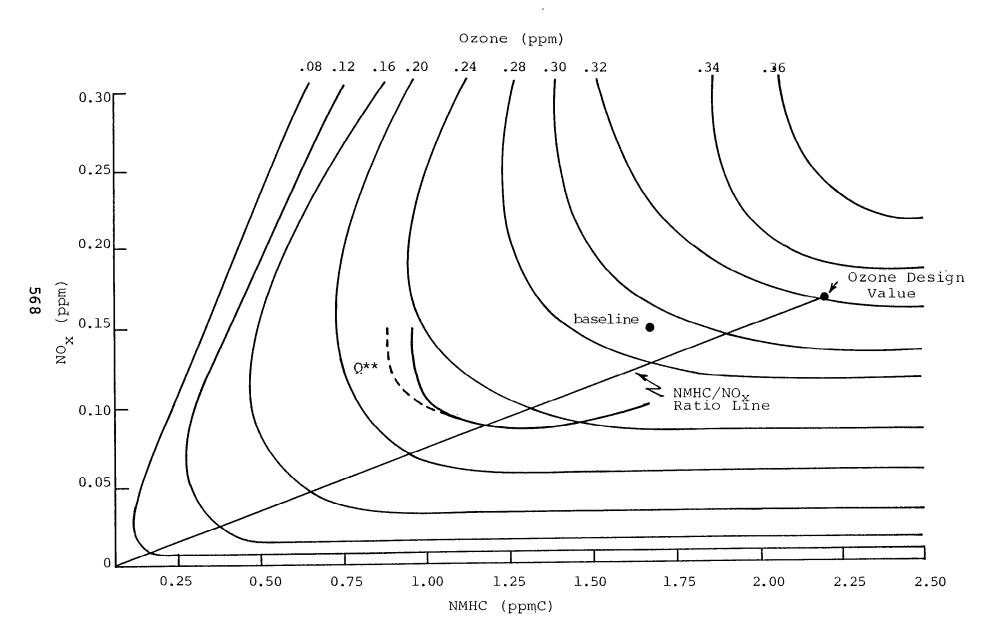


Figure A-3. Sensitivity Analysis of Least-cost Solution: Higher HC Prices

outward in the manner indicated in Figure a-3, making the initial steps to control hydrocarbons disproportionately cheaper. This might well have the effect of shifting the least-cost control strategy substantially toward hydrocarbon control, and away from $NO_{\mathbf{x}}$ control, to a point like \mathbf{Q}^{**} .

The cost of $\mathtt{NO}_{\mathbf{x}}$ abatement is sensitive to the treatment of damages averted other than effects on peak ozone concentrations. Reductions in $\mathtt{NO}_{\mathbf{x}}$ emissions have been shown to result in downwind abatement of a variety of other photochemical oxidation products, including nitrogen dioxide, nitric acid, PAN and sulfates. 28 These pollutants are implicated in crop damage, visibility impairment, acid precipitation, eye irritation and other effects. If the resource costs of NO_x abatement were credited with a monetary estimate of benefits from reduction of other damages, the net cost per ton of $\mathbf{NO}_{\mathbf{X}}$ control might be substantially reduced. While knowledge by which to quantify and value these secondary damages is lacking, conceptually the result is to shift outward the isocost line as indicated in Figure a-4. strategies are made relatively more attractive, since reductions in HC emissions, at least in the NY region, have been shown not to be as closely related to downwind concentrations of these secondary pollutants. It is apparent from Figure a-4 that the result might well be to shift the preferred control strategy sharply in the direction of $\mathtt{NO}_{\mathbf{x}}$ abatement, as indicated by a point like Q***.

The tendency of nonconvexity to produce all-or-nothing control strategies, therefore, is here muted somewhat by the

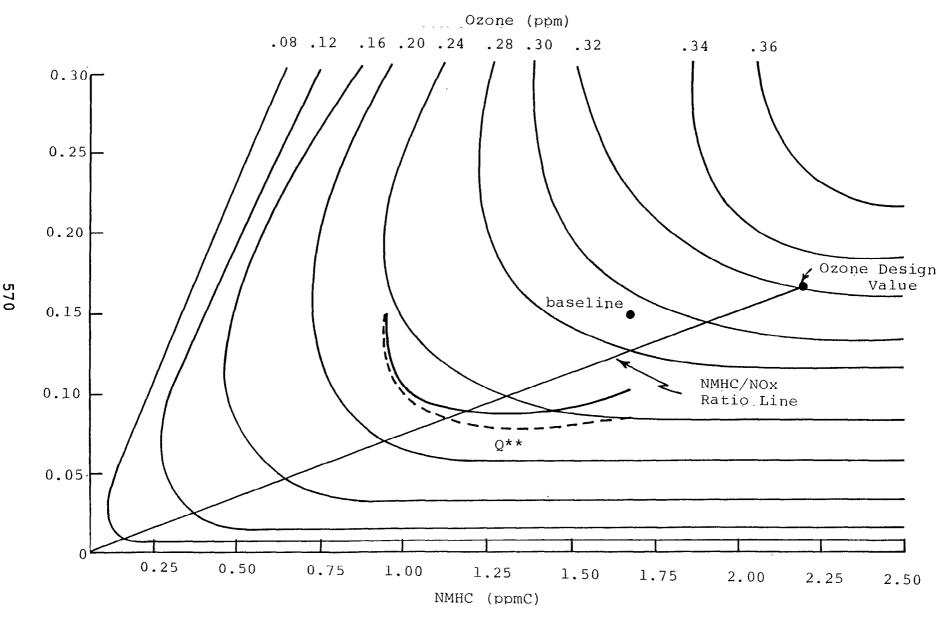


Figure A-4. Sensitivity of least-cost solution: credits for other effects of NO $_{\rm X}$ abatement

extreme convexity of abatement costs. However, least-cost control strategies remain highly sensitive to relatively small variations in either costs or effects. In that sense, knowledge of the convexity does not in this case greatly simplify the decision problem. However, it does simplify the process in one sense. Figure a-2 suggests an additional implication, that the burden of excess costs from a non-optimal or inefficient control strategy remain moderate even for wide deviations from the least-cost option. The price for choosing mostly ${\bf NO_X}$ batement strategy or a mostly HC strategy when a more balanced strategy would have been preferable is not too high, by comparison to other allocational problems not characterized by nonconvexities. In view of the extreme complexity of the decisions involved, this may provide some comfort to policy-makers.

NOTES

- 1. For air pollution, a useful recent critique is L. Lave and G. Omenn, <u>Clearing the Air</u>: <u>Reforming the Clean Air Act</u>, The Brookings Institution, Washington, D.C., 1981. Also <u>To Breathe Clean Air</u>, Report of the National Commission on Air Quality, Washington, D.C. 1981; Part 4.
- 2. Currently, due to Executive Order 12291, cost-benefit analysis is required in important regulatory decisionmaking. See "Federal Regulation," Executive Order 12291, Weekly Compilation of Presidential Documents, Vol. 17, (February 23, 1981), pp. 124130.
- 3. See, for example, D. Starrett, Fundamental Nonconvexities in "The Theory of Externalities," <u>Journal of Economic Theory</u>, IV, (1) February, 1972; 180-199; J.S. Winnick, "Convexity and Corner Solutions in the Theory of Externality," <u>Journal of Environmental Economics and Management</u>, IX (1), March, 1982; 29-41.
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- 5. For an economic analysis of the visibility problem, see R. Repetto, "The Economics of Visibility Protection: On a Clear Day You Can See a Policy," Natural Resources Journal, XXI (2), April, 1981: 355-370.
- 6. R. Zeckhauser and A. Fisher, "Averting Behavior and External Diseconomics" Discussion Paper, Kennedy School of Government, Harvard University, April, 1976: also, R. Porter and P. Courant, "Averting Expenditure and the Cost of Pollution," <u>Journal of Environmental Economies and Management</u>, VIII (4), December, 1981; 321329.
- 7. Council on Environmental Quality, <u>Annual Report</u>, <u>1980</u>, Washington, D.C., 1981.
- 8. To Breathe Clean Air, pp. 3429.
- 9. E.P.A., <u>Air Quality Criteria for Ozone and Other Photochemical Oxidants</u>, Vol. I, Washington, D.C., 1981.

- 10. K. Demerjian, J. Kerr, J. Calvert, "The Mechanisms of Photochemical Smog Formation," <u>Advances in Environmental Sciences and Technology</u>, N.Y., Wiley and Sons, 1974; National Research Council, National Academy of Sciences, <u>Ozone and Other Photochemical Oxidants</u>, Washington, DC., 1976.
- 11. W. Innes, "Effect of Nitrogen Oxide Emission on Ozone Levels in Metropolitan Regions," Environmental Science and Technology, XV (8), August, 1981; 904911. Wm. Glasson, "Effect of Hydrocarbon and NOx on Photochemical Smog Formation under Simulated Transport Conditions," Journal of the Air Pollution Control Association, XXX1 (11), November, 1981; 11691172.
- 12. M.C. Dodge, "Combined Use of Modeling Techniques and Smog Chamber Data to Derive OzonePrecursor Relationships," EPA, International Conference on Photochemical Oxidant Pollution and its Control, Proceedings, Research Triangle Park, N.C., 1977; B. Dimitriades, "An Alternative to the AppendixJ Method for Calculating Oxidant and NO2Related Control Requirements," ibid.
- 13. R. Dervent and O. Hov, "Computer Modeling Studies of the Impact of Vehicle Exhaust Emission Controls on Photochemical Air Pollution Formation in the U.K.," Environmental Science and Technology, VIII (9), Sept. 1975; 811826.
- 14. B. Dimitriades, "Oxidant Control Strategies: Urban Oxidant Control Strategy Derived from Existing Smog Chamber Data,"

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- 16. Report of the Photochemical Oxidant Panel to the National Commission on Air Quality, Washington, D.C., 1981 (mimeo).
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- 22. GCA, New York Metropolitan Region Study, Vol. 10; Bedford, MA.
- 23. Cleveland and Graedel, op. cit.
- 24. GCA, op. cit. Vol. IX
- 25. These inventories were compiled from data contained in the State Implementation Plans of the tri-state region, from EPA's National Emissions Data System, and from inventory data collected by GCA. For discussion of inventory quality, see GCA, op. cit., Vol. 2.
- 26. Cost-effectiveness data for NO_x abatement were supplied by James Wilson, <u>Energy and Environment Analysis</u>, <u>Inc.</u>, of Arlington, VA, Data on hydrocarbon control costs stems from GCA, <u>op. cit.</u>, Vols. IX and X.
- 27. The EKMA-OZIPP analysis of ozone formation was performed by GCA, and made available for this study. See GCA, op. cit. Vol. X.
- 28. Cleveland and Graedel, op. cit.